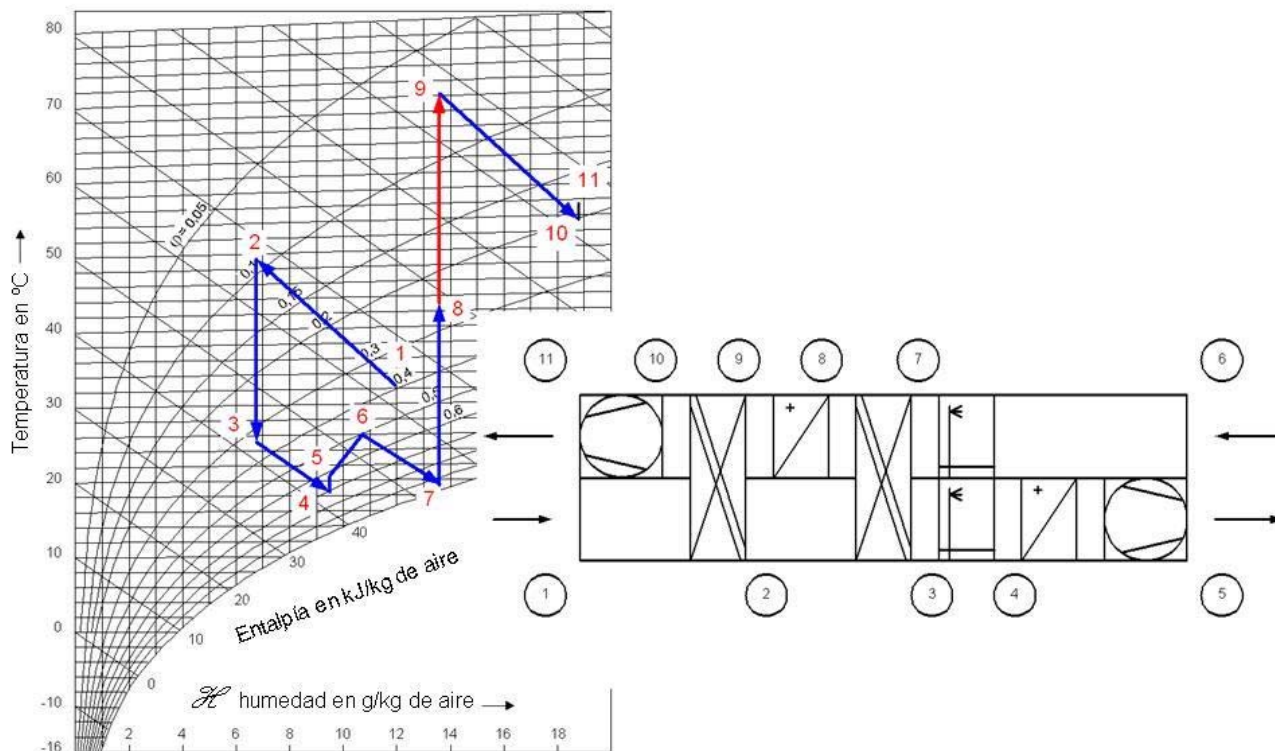


desiccant cooling system



Ciclo de refrigeración por adsorción utilizando energía térmica solar

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Adsorbedor

$$T_1 = 31 \text{ [C]}$$

$$P = 101,3 \text{ [kPa]}$$

$$x_1 = 0,4$$

$$h_1 = h \text{ ['AirH2O' ; } T = T_1 ; P = P ; R = x_1 \text{]}$$

$$w_1 = \omega \text{ ['AirH2O' ; } T = T_1 ; R = x_1 ; P = P \text{]}$$

$$T_1 = T_1$$

$$\dot{m}_1 = \frac{1}{2,198} \cdot 1 \text{ [kg/s]}$$

$$x_2 = 0,1$$

$$h \text{ ['AirH2O' ; } T = T_2 ; P = P ; R = x_2 \text{]} = h_1$$

$$w_2 = \omega \text{ ['AirH2O' ; } T = T_2 ; P = P ; R = x_2 \text{]}$$

$$T_2 = T_2 + 2 \text{ [C]}$$

$$h_2 = h \text{ ['AirH2O' ; } T = T_2 ; P = P ; w = w_2 \text{]}$$

$$w_2 = w_2$$

$$T_2 = T_2$$

Enfriador

$$w_3 = w_2$$

$$T_3 = 25 \text{ [C]}$$

$$h_3 = h \text{ ['AirH2O' ; } T = T_3 ; P = P ; w = w_3 \text{]}$$

$$Q_{\text{enf}} = [h_2 - h_3] \cdot \dot{m}_1$$

$$w_3 = w_3$$

$$T_3 = T_3$$

Enfriamiento por saturación parcial adiabática

$$T_4 = 18 \text{ [C]}$$

$$h_4 = h_3$$

$$h \text{ ['AirH2O' ; } T = T_4 ; P = P ; w = w_4 \text{]} = h_4$$

$$x_4 = \text{RH} \text{ ['AirH2O' ; } T = T_4 ; P = P ; w = w_4 \text{]}$$

$$Q_{\text{frío}} = [h_1 - h_4] \cdot \dot{m}_1$$

$$w_4 = w_4$$

$$T_4 = T_4$$

Variación de temperatura y humedad del aire en el local

$$T_6 = T_4 + 8 \text{ [C]}$$

$$x_6 = x_4 + 0,2$$

$$h_6 = h \text{ ['AirH2O' ; } T = T_6 ; P = P ; R = x_6 \text{]}$$

$$T_6 = \text{WB} \text{ ['AirH2O' ; } T = T_6 ; P = P ; R = x_6 \text{]}$$

$$w_6 = \omega \text{ ['AirH2O' ; } T = T_6 ; P = P ; R = x_6 \text{]}$$

$$T_6 = T_6$$

Enfriamiento por saturación parcial adiabática

$$h_7 = h_6$$

$$T_7 = T_6$$

$$h \text{ ['AirH2O' ; } T = T_7 ; P = P ; w = w_7 \text{]} = h_6$$

$$w_7 = w_7$$

$$T_7 = T_7$$

Intercambiador de calor

$$\dot{m}_1 \cdot [h_8 - h_7] = \dot{m}_1 \cdot [h_2 - h_3]$$

$$w_8 = w_7$$

$$h_8 = h ['AirH2O' ; T = T_8 ; P = P ; w = w_7]$$

$$\varepsilon = \frac{T_8 - T_7}{T_2 - T_3}$$

$$w_8 = w_8$$

$$T_8 = T_8$$

Calor de desorción

$$w_9 = w_8$$

$$T_9 = 75 \text{ [C]}$$

$$h_9 = h ['AirH2O' ; T = T_9 ; P = P ; w = w_9]$$

$$Q_{\text{solar}} = [h_9 - h_8] \cdot \dot{m}_1$$

$$w_9 = w_9$$

$$T_9 = T_9$$

Desorción

$$x_{10} = 0,18$$

$$h ['AirH2O' ; T = T_{10} ; P = P ; R = x_{10}] = h_9$$

$$T_{10} = T_{10} - 2 \text{ [C]}$$

$$w_{10} = \omega ['AirH2O' ; T = T_{10} ; P = P ; R = x_{10}]$$

$$h_{10} = h ['AirH2O' ; T = T_{10} ; P = P ; w = w_{10}]$$

$$w_{10} = w_{10}$$

$$T_{10} = T_{10}$$

Rendimiento global

$$\text{COP} = \frac{Q_{\text{frío}}}{Q_{\text{solar}}}$$

SOLUTION

Unit Settings: [kJ]/[C]/[kPa]/[kg]/[degrees]

$$\text{COP} = 0,647$$

$$h_1 = 60,0 \text{ [kJ/kg]}$$

$$h_2 = 62,0 \text{ [kJ/kg]}$$

$$h_4 = 40,0 \text{ [kJ/kg]}$$

$$h_7 = 73,4 \text{ [kJ/kg]}$$

$$h_9 = 126,3 \text{ [kJ/kg]}$$

$$P = 101,3 \text{ [kPa]}$$

$$Q_{\text{frío}} = 9,08 \text{ [kW]}$$

$$T_{10} = 61,55 \text{ [C]}$$

$$T_6 = 24,32 \text{ [C]}$$

$$T_2 = 46,61 \text{ [C]}$$

$$T_4 = 18 \text{ [C]}$$

$$T_7 = 24,32 \text{ [C]}$$

$$T_9 = 75 \text{ [C]}$$

$$T_1 = 31,0 \text{ [C]}$$

$$T_2 = 46,6 \text{ [C]}$$

$$T_4 = 18,0 \text{ [C]}$$

$$T_7 = 24,3 \text{ [C]}$$

$$T_9 = 75,0 \text{ [C]}$$

$$w_2 = 0,005826$$

$$w_4 = 0,008636$$

$$w_7 = 0,01923$$

$$w_9 = 0,01923$$

$$w_{10} = 0,02235$$

$$w_3 = 0,005826$$

$$w_7 = 0,01923$$

$$w_9 = 0,01923$$

$$x_1 = 0,400$$

$$x_4 = 0,672$$

$$\varepsilon = 0,976$$

$$h_{10} = 118,3 \text{ [kJ/kg]}$$

$$h_3 = 40,0 \text{ [kJ/kg]}$$

$$h_6 = 73,4 \text{ [kJ/kg]}$$

$$h_8 = 95,4 \text{ [kJ/kg]}$$

$$\dot{m}_1 = 0,45496 \text{ [kg/s]}$$

$$Q_{\text{enf}} = 10,01 \text{ [kW]}$$

$$Q_{\text{solar}} = 14,04 \text{ [kW]}$$

$$T_2 = 44,61 \text{ [C]}$$

$$T_1 = 31 \text{ [C]}$$

$$T_3 = 25 \text{ [C]}$$

$$T_6 = 26 \text{ [C]}$$

$$T_8 = 45,41 \text{ [C]}$$

$$T_{10} = 59,55 \text{ [C]}$$

$$T_{10} = 59,55 \text{ [C]}$$

$$T_3 = 25,0 \text{ [C]}$$

$$T_6 = 26,0 \text{ [C]}$$

$$T_8 = 45,4 \text{ [C]}$$

$$w_1 = 0,01124$$

$$w_3 = 0,005826$$

$$w_6 = 0,01854$$

$$w_8 = 0,01923$$

$$w_{10} = 0,02235$$

$$w_2 = 0,005826$$

$$w_4 = 0,008636$$

$$w_8 = 0,01923$$

$$x_2 = 0,1$$

$$x_{10} = 0,18$$

$$x_6 = 0,872$$

No unit problems were detected.

Arrays Table

	w_i	T_i [C]
1	0,01124	31
2	0,005826	46,61
3	0,005826	25
4	0,008636	18
5		
6	0,01854	26
7	0,01923	24,32
8	0,01923	45,41
9	0,01923	75
10	0,02235	59,55

